

Electrochemical oxidation of t-CNSL using selective cathodic and anodic strategies to simultaneously produce green hydrogen and acetic acid

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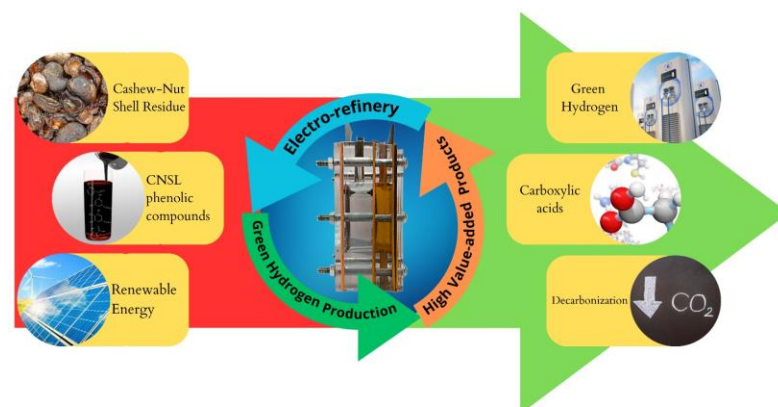
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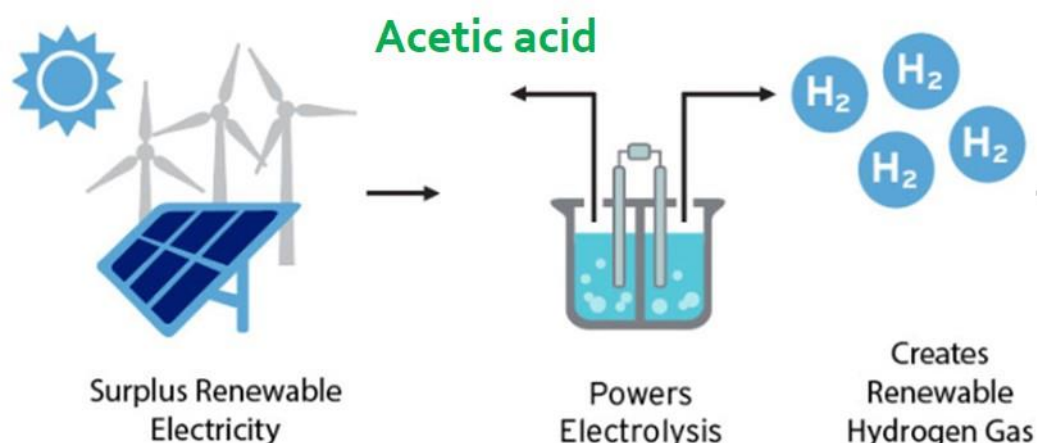
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Introduction

In this study, the scale-up of electrochemical treatment was investigated by treating technical cashew nutshell liquid (t-CNSL) effluent using a divided-electrochemical reactor photovoltaically-driven, for the first time, with a boron-doped diamond (BDD) electrode as the anode and a Ni-Fe-based SS (stainless steel) mesh as the cathode. The results clearly demonstrated that a divided electrochemical reactor was successfully scaled up, favoring a selective accumulation of the acetic acid concentration (600 mg L⁻¹). Additionally, simultaneous production of green H₂ (0.73 L) was achieved when the biomass effluent of 0.10% t-CNSL in 1 mol L⁻¹ NaOH was electrolyzed by applying 40 mA cm⁻².



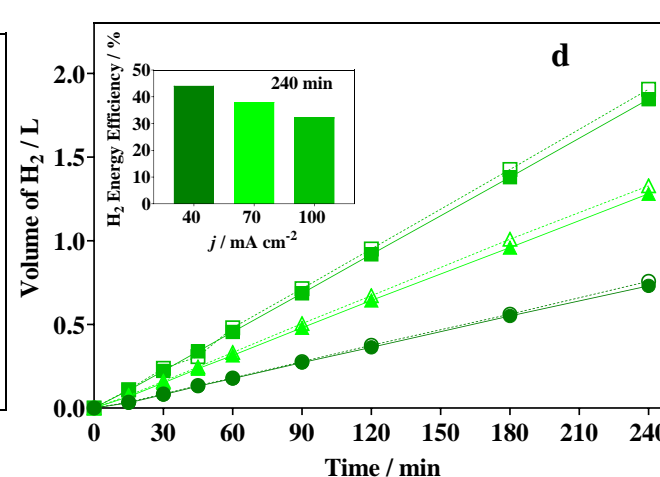
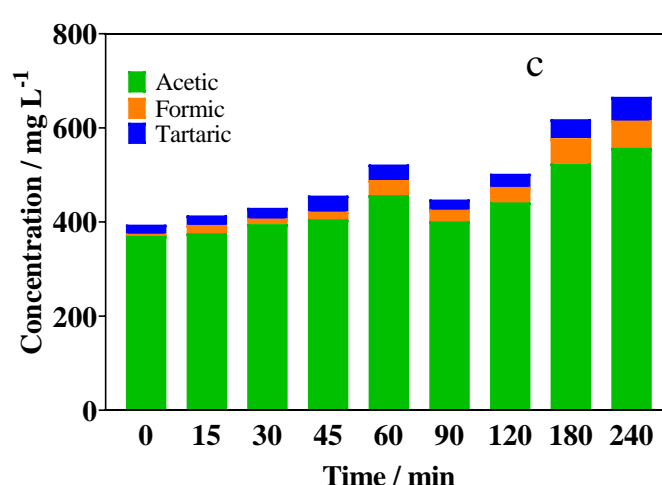
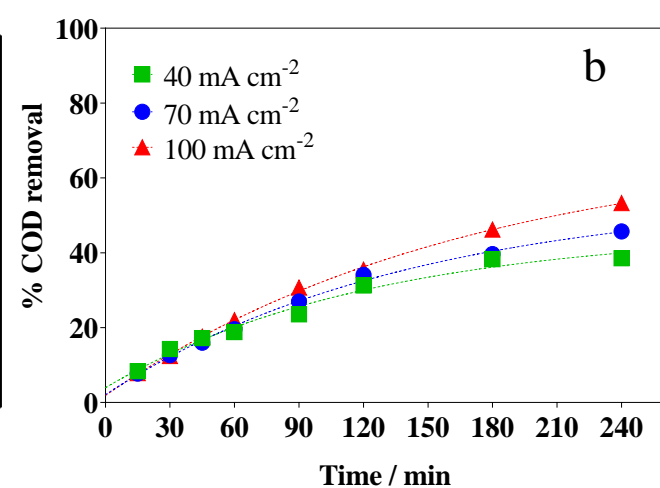
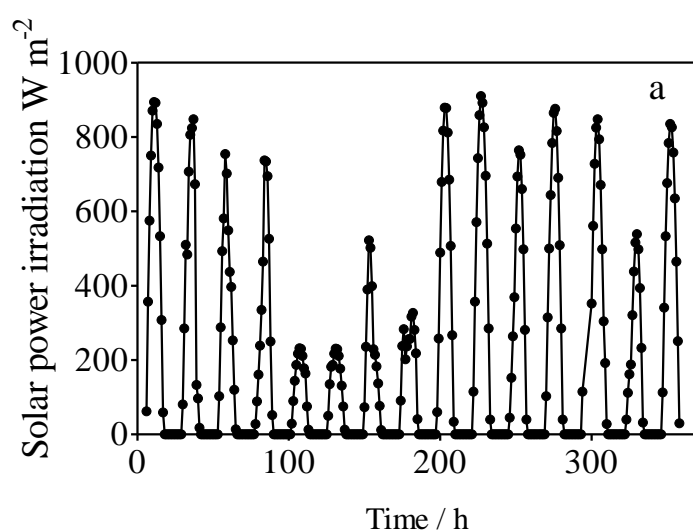
Methodology



t-CNSL was acquired from a cashew nut processing plant located in Mossoró, Rio Grande do Norte, Brazil. The experiments were conducted using galvanostatic conditions during 240 min. AEM-type E-reactor, in a sandwich configuration, consists of an anodic and cathodic compartment, separated by an AEM with a BDD (10 cm²) as anode and Ni-Fe based SS mesh as cathode

Results

Electrochemical treatment of 0.01% t-CNSL in 1.0 mol L⁻¹ NaOH removed 38.54% of organic matter with 40 mA cm⁻². Using 0.05% t-CNSL, the concentrations of acetic, tartaric and formic acids were 565.3, 41.2 and 61.2 mg L⁻¹ after 240 min of electrolysis. Green H₂ production was linear over time, with volumes of 0.73 L, 1.28 L and 1.84 L for 40, 70 and 100 mA cm⁻², respectively, and energy efficiencies of 44%, 38 % and 33%.



Conclusion

In conclusion, this work has demonstrated an effective strategy for energy-saving electrochemical H₂ production by replacing the conventional water splitting with the electrochemical treatment of biomass effluent. Overall, our work presents a promising approach to enhance the cost-effectiveness of the electrochemical H₂ production by reducing the electrolysis voltage and simultaneously biomass upgrading obtaining high value-added products (like acetic acid).

